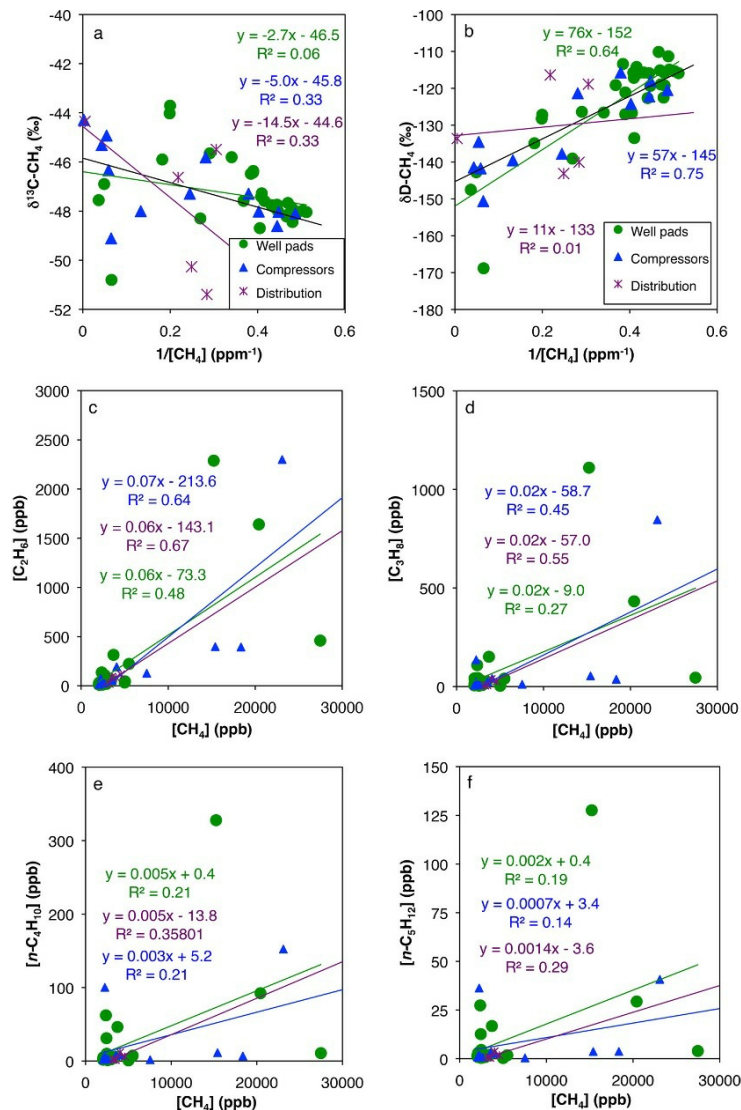


## (84-240513-C) Hydrogen Isotopes as Tracers of Methane Sources

A. Townsend-Small

School of Environment and Sustainability, University of Cincinnati; 949-614-6250, E-mail: townseay@ucmail.uc.edu

Scientists investigating rising  $\text{CH}_4$  concentrations regionally and globally have relied on a variety of approaches to estimate sources of excess methane, including inventories of sources, carbon isotopes of  $\text{CH}_4$ , and methane:ethane ratios. In this presentation, I will give examples from several studies that have measured the hydrogen stable isotopic composition ( $\delta\text{D}$ ) of  $\text{CH}_4$  in both sources (i.e., at the "bottom up" level) and in well mixed air masses (for "top down" source apportionment), and argue that this tracer is a valuable potential tracer of global methane sources. Advantages of hydrogen isotopes include 1), consistent  $\delta\text{D}$  ratios of  $\text{CH}_4$  within oil and gas basins as compared to  $\delta^{13}\text{C}$  and  $\text{CH}_4:\text{C}_2\text{H}_6$ ; 2), most sources have a distinct  $\delta\text{D}-\text{CH}_4$  from atmospheric background, which makes it easier to distinguish small enhancements in  $\text{CH}_4$ , unlike  $\delta^{13}\text{C}$ , where some oil and gas sources have similar signatures to background air; and 3), the ability to use a two-endmember mixing model for source apportionment rather than a one-endmember mixing model, which is the case with  $\text{CH}_4:\text{C}_2\text{H}_6$  (because biogenic sources do not have  $\text{C}_2\text{H}_6$ ). Some disadvantages of using  $\delta\text{D}$  vs the others include 1), there are no in situ instruments available for measuring  $\delta\text{D}$ , as there are for  $\delta^{13}\text{C}$  and  $\text{C}_2\text{H}_6$ , and fewer laboratories measuring this isotope in  $\text{CH}_4$ ; and 2), currently there are somewhat larger sample volume requirements for  $\delta\text{D}$  than  $\delta^{13}\text{C}$ , although still much smaller than in the recent past.



**Figure 1.** Composition of methane from natural gas sources in the Barnett Shale region. (a) Keeling plot of  $\delta^{13}\text{C}-\text{CH}_4$  vs  $1/[\text{CH}_4]$ ; (b)  $\delta\text{D}-\text{CH}_4$  vs  $1/[\text{CH}_4]$ ; (c)  $[\text{C}_2\text{H}_6]$  vs  $[\text{CH}_4]$ ; (d)  $[\text{C}_3\text{H}_8]$  vs  $[\text{CH}_4]$ ; (e)  $[\text{n}-\text{C}_4\text{H}_{10}]$  vs  $[\text{CH}_4]$ ; and (f)  $[\text{n}-\text{C}_5\text{H}_{12}]$  vs  $[\text{CH}_4]$ .