



# Large Scale Temporal and Spatial Gradients of Carbon Dioxide as Derived from the NOAA/ESRL Aircraft Profiles



S.E. Peterson<sup>1</sup>, C. Sweeney<sup>1</sup>, E.J. Dlugokencky<sup>2</sup>, D.W. Guenther<sup>1</sup>, K.A. Masarie<sup>2</sup>, P.M. Lang<sup>2</sup>, and M.J. Heller<sup>1</sup>

<sup>1</sup>Cooperative Institute for Research in Environmental Sciences, University of Colorado, UCB 216, Boulder, CO 80309-0216; Sarah.Peterson@noaa.gov  
<sup>2</sup>NOAA Earth System Research Laboratory, 325 Broadway, Boulder, CO 80305

## 1. Abstract

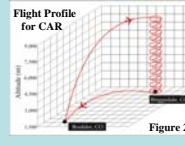
To assess the impacts of fossil fuel combustion on atmospheric composition and climate, Dave Keeling began measurements of carbon dioxide (CO<sub>2</sub>) at the Mauna Loa Observatory in 1957. While such measurements are still useful in quantifying the atmospheric burden of CO<sub>2</sub>, other measurement strategies are necessary to determine the fate and source of fossil-derived CO<sub>2</sub>. To address this need the NOAA/ESRL GMD Carbon Cycle Greenhouse Gases Group created the aircraft sampling network in 1992. Currently this network consists of 20 aircraft sites primarily located in North America. The air samples collected are analyzed in Boulder, Colorado for CO<sub>2</sub>, carbon monoxide (CO), hydrogen (H<sub>2</sub>), nitrous oxide (N<sub>2</sub>O), sulfur hexafluoride (SF<sub>6</sub>), and methane (CH<sub>4</sub>) as well as halocarbons and the isotopic ratio of CO<sub>2</sub>. In this study we examine the spatial and temporal variability of CO<sub>2</sub> throughout a climatological year.

## 2. Sampling Equipment



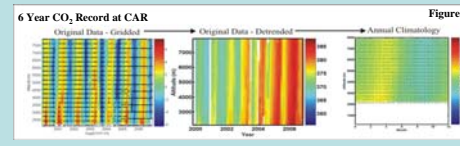
Air samples are collected using two different instruments, the programmable flask package (FPF) and the programmable compressor package (PCP) (Figure 1). Each FPF holds twelve 0.75L glass flasks connected by a stainless steel welded manifold and a pressure sensor. The PCP contains the systems power, pumps, flow meter, and electronics. These two units are deployed on small aircrafts for programmable time-interval or event based sampling of atmospheric CO<sub>2</sub> and other trace gases.

## 3. Aircraft Profiles



This program collects discrete air samples from 500-8,000 m above ground. Figure 2 shows the aircraft profile taken at Briggsdale, CO (CAR). At present, 20 aircraft sites are being sampled every two to three weeks as a part of the North American Carbon Program. These samples are collected using single and twin engine Cessnas. Trace gases from the FPFs are measured at ESRL.

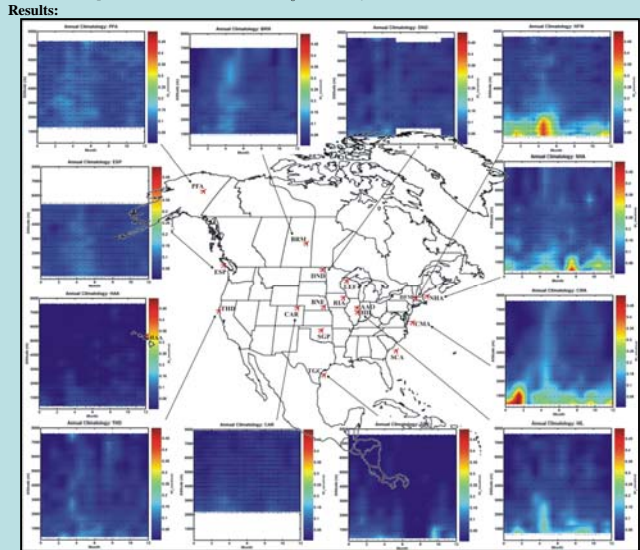
## 4. Climatology Methods



This study uses CO<sub>2</sub>, CO, and SF<sub>6</sub> data collected from the NOAA/ESRL Aircraft Network. All of the data used in this study were binned by month, detrended, and then adjusted to one common year (2004) by normalizing them with the Mauna Loa surface site measurements. Figure 3 shows this process for the sampling record at CAR. The black dots show the sampling densities that have been used to create the annual climatology.

## 5. Sulfur Hexafluoride Data

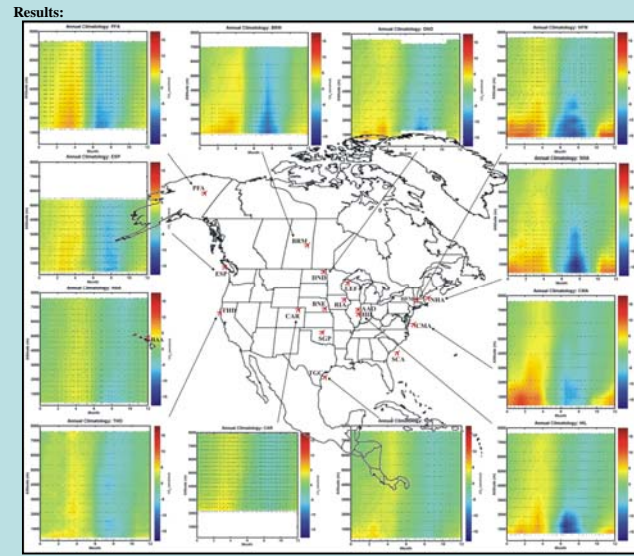
**Background:** SF<sub>6</sub> is an important greenhouse gas to monitor due to its increasing atmospheric burden, high global warming potential (GWP), and long lifetime. During the 1980s and 1990s surface measurements show that SF<sub>6</sub> was increasing at a rate of 7% per year (Geller *et al.*, 1997; Maiss and Brenninkmeijer, 1998). SF<sub>6</sub> by mass, is 23,000 times more effective at changing radiative forcing than CO<sub>2</sub> over a 100 year time scale (IPCC, 2001). The lifetime of SF<sub>6</sub> is 3,200 years (Ravishankara *et al.*, 1993). It is primarily used to insulate high voltage electrical equipment such as electrical transmission and distribution systems in gas-insulated switching gear transformers. The major sink of SF<sub>6</sub> is a chemical reaction in the mesosphere. There is little uptake from the ocean because SF<sub>6</sub> is relatively insoluble in water.



**Conclusions:** Unlike CO<sub>2</sub> and CO gradients which reflect vertical mixing, seasonal and interannual changes in interhemispheric transport, and sources and sinks, SF<sub>6</sub> is relatively constant. Figure 4 shows high levels of SF<sub>6</sub> in three East Coast sites: Harvard Forest, Massachusetts (HFM); Worcester, Massachusetts (NHA); and Cape May, New Jersey (CMA). The elevated levels seen at the East Coast sites are likely attributed to the close proximity to SF<sub>6</sub> sources. During summer, enhanced vertical mixing dilutes SF<sub>6</sub> signals in the planetary boundary layer (PBL) with SF<sub>6</sub>-depleted air from the free troposphere (FT). This occurs in the Northern Hemisphere (NH), reducing the quantity of SF<sub>6</sub> in the PBL (Gloor *et al.*, 2007). In general the mixing ratio of SF<sub>6</sub> at most sites show a negative vertical gradient (Gloor *et al.*, 2007). This is clearly shown in Figure 4, especially on the Eastern Coast where there are higher levels of background SF<sub>6</sub>.

## 6. Carbon Dioxide Data

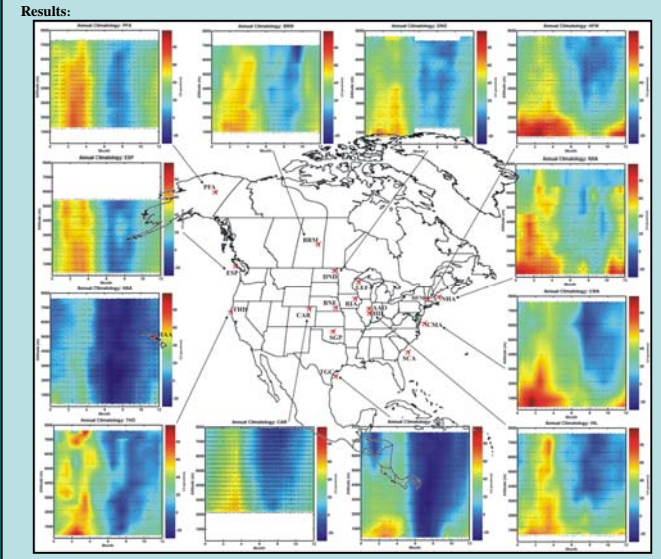
**Background:** CO<sub>2</sub> is a major component of the carbon cycle. The annual mean global average growth rate of CO<sub>2</sub> from 1990 to 1999 was 3.2±0.1 PgC/yr (IPCC, 2001). CO<sub>2</sub> is released into the atmosphere when carbon-containing fossil fuels such as oil, natural gas, and coal are burned. This process accounts for ~75% of CO<sub>2</sub> emissions (IPCC, 2001). The rest of the emissions are due to changes in land use. Long-term natural sinks of CO<sub>2</sub> include ocean uptake as well as a build up of terrestrial biomass. In the short-term, the seasonal cycle at each site is driven primarily by summertime uptake of CO<sub>2</sub> (photosynthesis) and the wintertime outgassing of CO<sub>2</sub> (respiration).



**Conclusions:** In the NH we see the largest uptake of CO<sub>2</sub> in the Midwest due to crop production and on the East Coast due to deciduous forests. During the summer months we see that vegetation uptake of CO<sub>2</sub> exceeds vegetation respiration and fossil fuel emissions. This leads to lower CO<sub>2</sub> in the PBL than in the FT. The opposite occurs in the winter when fossil fuel emissions and vegetation respiration become dominant. As air masses move from west to east the concentration differences are amplified leading to larger differences between BL and FT CO<sub>2</sub> concentrations. The West Coast has weaker horizontal CO<sub>2</sub> fluxes and exhibits a less defined seasonal cycle primarily due to the influx of marine air from westerly winds. By examining the plots in Figure 5 we can see that the sources and sinks are more defined and the vertical mixing is less defined on the East Coast.

## 7. Carbon Monoxide Data

**Background:** CO is a very weak greenhouse gas, but its reaction with hydroxyl radical, OH, affects the lifetimes of some long-lived greenhouse gases like CH<sub>4</sub>. The primary anthropogenic source of CO is the incomplete combustion of carbon-containing fuels. Total emissions of CO are ~2800 Tg(CO)/yr with approximately half from *in situ* production and half from direct emissions (IPCC, 2001). Through natural processes in the atmosphere, CO is eventually oxidized to CO<sub>2</sub>.



**Conclusions:** Because CO has an atmospheric lifetime of a few months, it is spatially and temporally more variable than longer-lived gases. A combination of vertical mixing and the seasonally varying concentration of OH drives the large seasonal cycle in CO, which is seen at all of our sampling sites in North America (Figure 6). This cycle is seen throughout the atmospheric column and can be attributed to mid-latitude boundary layer build up of CO. This column is carried north and around the Arctic Circle only to be entrained in the FT and mixed down into the PBL of sites like Poker Flat, Alaska (PFA).

## 8. Conclusions

This study shows that there are similarities in the profiles of SF<sub>6</sub>, CO<sub>2</sub>, and CO. Each of these gases see enhancements throughout the atmospheric column as air masses move from west to east across North America. The largest enhancements are seen in the boundary layer at East Coast sites for all three gases. We see stronger seasonality throughout the column for CO than CO<sub>2</sub>, because the sink for CO is active throughout the troposphere while the sink for CO<sub>2</sub> is only at the surface. Even without detailed information on the concentration and distribution of OH, we can make a reasonable conclusion that fossil fuel emissions are affecting the seasonal cycles of CO and CO<sub>2</sub>. Because emissions of SF<sub>6</sub> are fairly well known, it is an excellent tracer for testing transport in models used to determine the fate of fossil fuel CO<sub>2</sub> (e.g., TM5).

## 9. References

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## 10. Acknowledgments

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