

WHAT'S SO SPECIAL ABOUT CO2?

OBJECTIVE: Students will:

- ♣ Understand the concept of resonance frequency
- ♣ Simulate the resonance frequency of certain molecules using models
 with different characteristics.
- Analyze data

MATERIALS:

- ♣ 12 Styrofoam balls about 2-3 inches in diameter
- 4 5 3 ft. lengths of thin, springy steel rods, 1/16 in in diameter
- 4 clocks/watches with a second hand
- Pliers or wire cutters
- ♣ Paper/pencil

PREPARATION:

- 1. Before class, cut the steel rods into 1 foot lengths.
- 2. Color or label the balls to represent C, H or O.
- 3. Prepare copies of the molecule information card for distribution.
- 4. In class:
 - Divide the class into 4 teams.
 - ♣ Give each team the appropriate molecule information card.

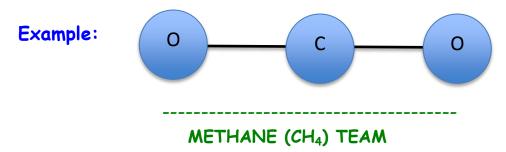
 (CO₂, CH₄, O₂, N2)
 - ♣ Each team will construct a model of a molecule.
- 5. When all the models are completed, teams should exchange models so that each group can compare how the different molecules respond to different frequencies of vibration.
- 6. Be sure that students complete their **DATA TABLE** and the **CONCLUSIONS** section at the end of the activity.

CARBON DIOXIDE (CO2) TEAM

MATERIALS: 3 Styrofoam balls, 1 3ft. long steel rod, 1 clock/watch

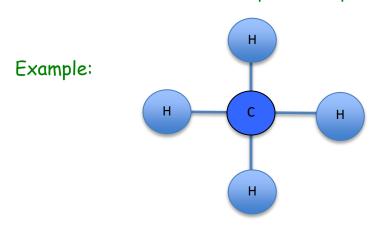
PROCEDURE:

- Insert the metal rod through a ball representing a carbon atom.
- Adjust the rod so that it goes directly into the center of the ball.
- ♣ Add a ball to each end of the rod to represent oxygen.



MATERIALS: 5 Styrofoam balls,2 3ft. long steel rods, 1 clock/watch; **PROCEDURE**:

- Insert 1, 3 feet long rod through a ball representing a carbon atom;
- ♣ Insert a second 3 ft. long rod through the carbon atom at a right angle to the first rod. You should now have a carbon atom with 4 "spokes."
- ♣ Add 1 ball to the end of each spoke to represent H atoms.



OXYGEN (O2) TEAM

MATERIALS: 2 Styrofoam balls, 2 short (1foot) steel rods, clock/watch

PROCEDURE:

- Insert 2 short rods into one of the balls representing an O atom.
- ♣ Push the second ball onto the ends of the 2 rods so the 2 "atoms" are joined by the rods.

EXAMPLE:



NITROGEN (N2) TEAM

MATERIALS: 2 Styrofoam balls, 3 short (1 foot) rods, clock/watch:

PROCEDURE:

- Insert 3 short rods into one side of a ball representing a N atom. Make the rods parallel.
- Push the second ball onto the other ends of the rods so the two "atoms" are joined by the rods.

EXAMPLE:

