## Measurement of VOCs in Marine Air at Cape Grim using PTR-MS

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Sarah Lawson , Ian Galbally, Erin Dunne, John Gras and Paul Krummel ESRL Global Monitoring Annual Conference, 18-19 May 2010



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#### Outline



- Measuring VOCs using PTR-MS
  - Introduction to PTR-MS as a measurement technique
  - Review of suitability of PTR-MS for measuring VOCs in background air
  - How we use PTR-MS at Cape Grim
- Results from Cape Grim
  - VOC concentrations from two recent campaigns and comparison with other remote oceanic sites
  - Diurnal cycles of VOCs
  - VOCs from nearby biomass burning
- Plans for future work





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#### Why measure VOCs at Cape Grim?



- VOCs,  $NO_x$  and sunlight drive atmospheric chemistry
  - Essential for ozone production
- VOCs are precursors of Secondary Organic Aerosol (SOA) marine sources
  - High proportion of aerosol in atmosphere is thought to be SOA but formation mechanisms and properties poorly understood
  - SOA climatically active –uncertainty in radiative forcing attributed to aerosol is very high



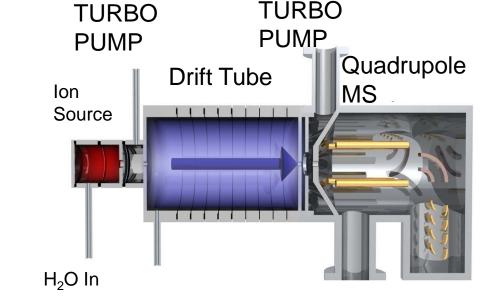


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#### **Background on the PROTON TRANSFER REACTION MASS SPECTROMETER**



- Real time measurements, no Carrier gas, and no consequent dilution
- Ion Source > 98% H<sub>3</sub>O<sup>+</sup>, <2% O<sub>2</sub><sup>+</sup>
- Primary reaction is: VOCH<sup>+</sup> + H<sub>2</sub>O  $VOC + H_3O^+$
- detects compounds with proton affinities (PA) greater than H<sub>2</sub>O
- **Compounds detected include:** organics containing O, N and S, alkenes and aromatics
- Cannot differentiate compounds with same MW



Air Sample In



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#### Review of suitability of PTR-MS to measure VOCs

| VOC                            | Protonated<br>Mass | OK in<br>background<br>air?            | Issues at this mass   | Inter-comparison studies  |  |
|--------------------------------|--------------------|--|---|---|--|
| methanol                       | 33                 | Y                                      | -   | FTIR (Christian et al., 2004) (Karl et al.,<br>2007)GCMS (de Gouw et al., 2003)   |  |
| acetonitrile                   | 42                 | Y                                      | O <sub>2</sub> <sup>+</sup> + propene but not significant<br>when propene <100 ppt  | GC-MS (de Gouw et al., 2003)  |  |
| ethanol                        | 47                 | Challenging                            | Formic acid<br>Low sensitivity to ethanol due to<br>fragmentation to H <sub>3</sub> O <sup>+</sup> (Inomata<br>Tanimoto, 2009b)<br>Di methyl ether? | Saphir (Apel et al., 2008)  |  |
| acetone                        | 59                 | Y                                      | Propanal contributes (~10%)<br>possibly glyoxal   | GC-MS (de Gouw et al., 2003)<br>(Karl et al., 2007)<br>Saphir (Apel et al., 2008) |  |
| di methyl<br>sulphide<br>(DMS) | 63                 | Y                                      | Hydrated acetaldehyde may<br>contribute where [acetaldehyde] is<br>>> [DMS]   | GCMS (de Gouw Warneke, 2007)  |  |
| isoprene                       | 69                 | Y (providing<br>no biomass<br>burning) | Furan<br>C₅H <sub>8</sub> hydrocarbons  | GCMS(de Gouw et al., 2003) (Karl et al.,<br>2007)                                 |  |
| terpenes                       | 81, 137            | Y                                      | -   | GC-MS (de Gouw et al., 2003)  |  |
| acetaldehyde                   | 45                 | Y                                      | Ethylene oxide?   | GC-MS (de Gouw et al., 2003)  |  |





#### How we use PTR-MS at Cape Grim



- Daily zeros and calibrations with gas standards
- Calibrated concentrations (5) minute) for a range of hydrocarbons and oxygenated VOCs
- Presented here is data from February 2006 and December 2007
- Data selected for 'clean air' conditions, CN10<3000 cm<sup>-3</sup>, radon <200 SCM



Radon data provided by Wlodek Zahorowski



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# Results – VOC concentrations (ppt) at Cape Grim and other Oceanic sites



| Protonated<br>Mass | Most Probable<br>Compound | Lifetime in<br>marine<br>boundary<br>layer<br>(days) | Cape<br>Grim<br>2007 | Cape<br>Grim<br>2006 [1] | Cape Grim<br>1988-1993<br>[2] | Southern<br>Hemisphere<br>Mid<br>Latitude<br>Oceanic [3]<br>[4] | Tropical<br>Oceanic<br>[5][6][7] |
|--------------------|---------------------------|--|----------------------|--------------------------|-------------------------------|---|----------------------------------|
| 33                 | Methanol                  | 15   | 633                  | 476                      | -                             | 546   | 575-890                          |
| 42                 | Acetonitrile              | 471  | 32                   | 25                       | -                             | 20  | 111-142                          |
| 45                 | Acetaldehyde              | 0.9  | 53                   | nd                       | -                             | 290   | 204                              |
| 59                 | Acetone                   | 66   | 61                   | 118                      | -                             | 127 - 450   | 466-530                          |
| 63                 | DMS                       | 1.5  | 95                   | ~80                      | 110                           | 77  | 50-89                            |
| 69                 | Isoprene                  | 0.1  | 21                   | 14                       | -                             | 40 - 66   | -                                |
| 81                 | monoterpenes              | ~0.1   | 25                   | nd                       | -                             | 10  | -                                |

1. Galbally, I.E., et al., Environmental Chemistry, 2007. 4(3): p. 178-182.

2. Ayres, G.P. and R.W. Gillett, Journal of Sea Research, 2000. 43: p. 275-286.

3.Colomb, A., et al., Environmental Chemistry, 2009. 6(1): p. 70-82.

4. Williams, J., et al., Environmental Chemistry, 2010 7(2): p. 171-182.

5. Warneke, C. and J.A. de Gouw, Atmospheric Environment, 2001. 35(34): p. 5923-5933.

6 .Singh, H.B., et al., Journal of Gephysical Research, 2004. 109.

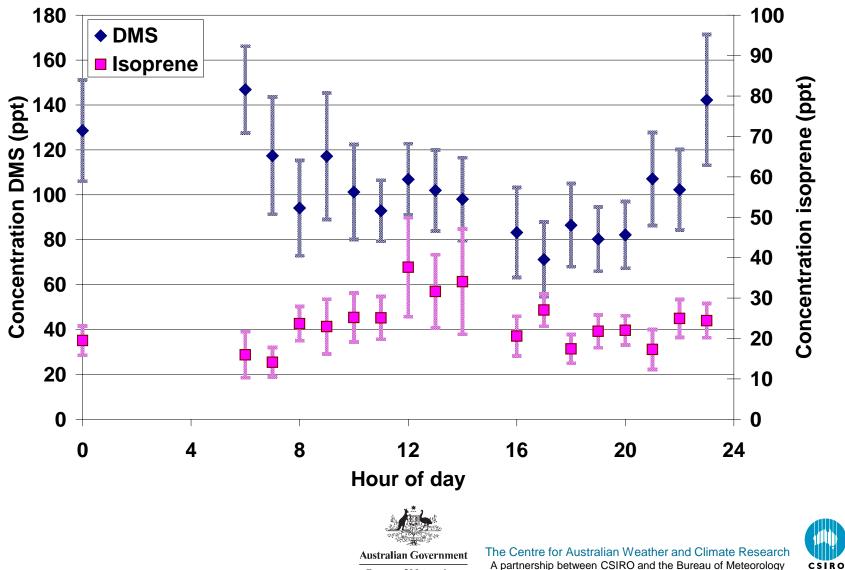
7. Williams, J., et al., Geophysical Research Letters, 2004. 31(23); p. 5.



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#### DMS and Isoprene diurnal cycles in clean air Dec 2007



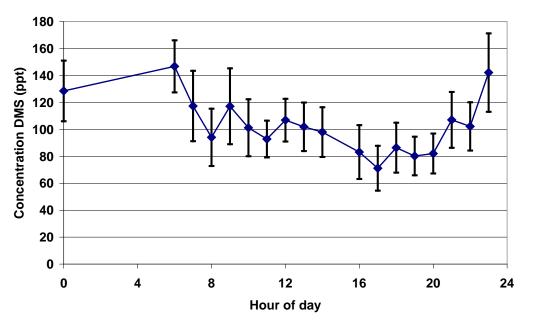
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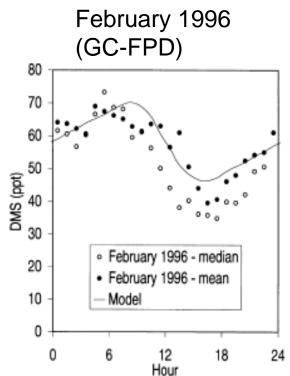
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#### DMS diurnal cycle – past and present



10 days in December 2007 (PTR-MS)





From: Ayers and Gillett, Journal of Sea Research 43 (2000) 275–286

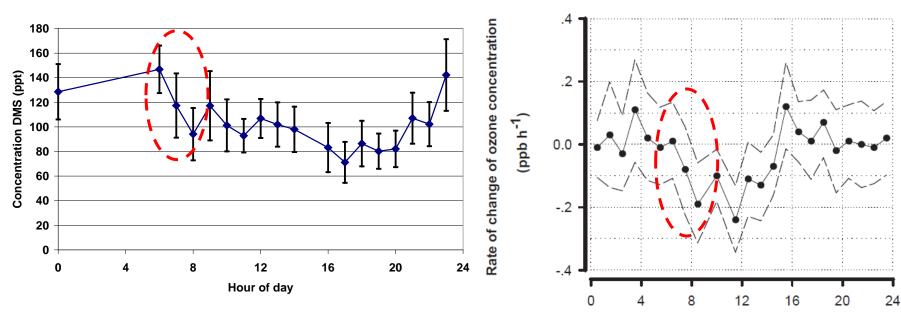


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#### Halogen chemistry?

DMS Dec 2007



Hour of day

O<sub>3</sub> Dec-Jan 1985-1997

From: Galbally et al, Geophysical Research Letters, 27 (2000), 3841-3844



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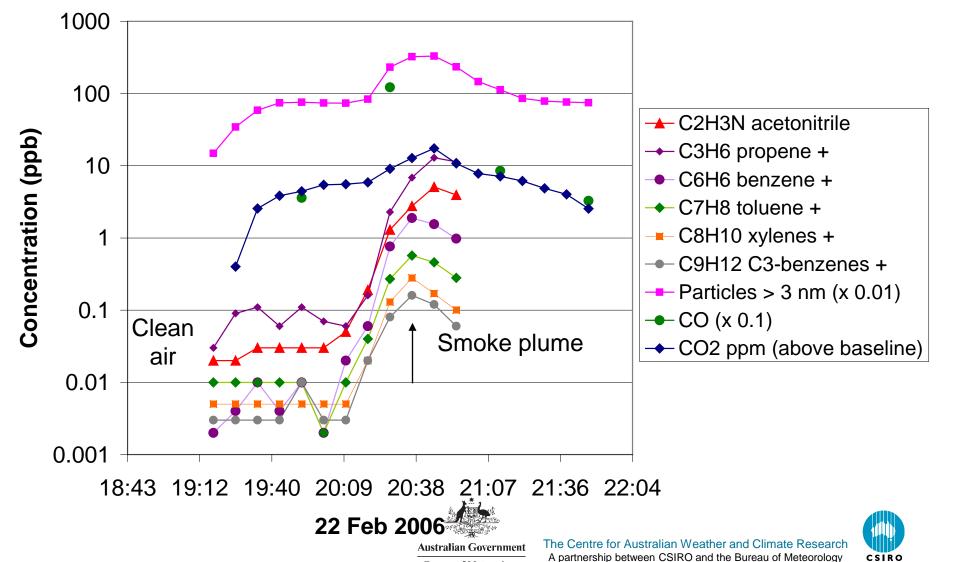
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#### Local Biomass Burning -Smoke plume from Robbins Island fire 22 Feb 2006



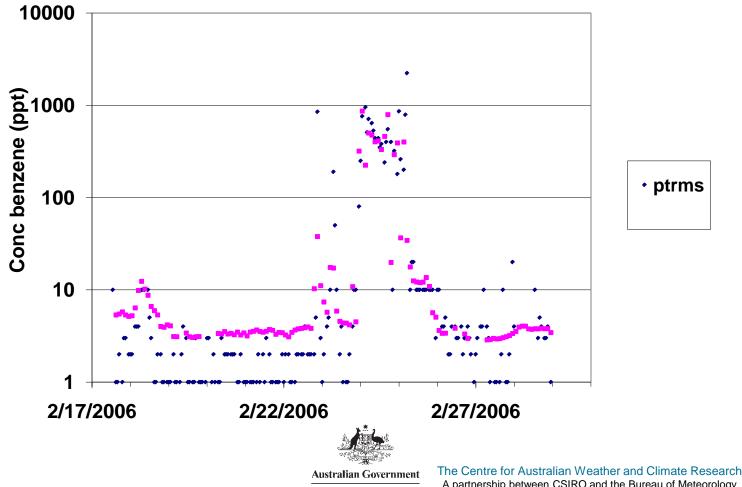
#### Hydrocarbons and indicators in the smoke plume





#### Benzene in smoke plume measured by PTR-MS and Medusa GC-MS

#### PTRMS (1 hour average) and Medusa (20 minute average)



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#### Future Work



- Calibration of AGAGE Medusa GC-MS with VOC standard to extend range
- Future campaigns with
  - particle sizing aerosol, PTR-MS side by side
  - Inlet selective scrubbers on PTR-MS eg acid scrubber (formic & acetic)
  - Switchable Reagent Ion (NO<sup>+</sup>,  $O_2^+$ ,  $H_3O^+$ ) for better separation of species





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#### Sarah Lawson

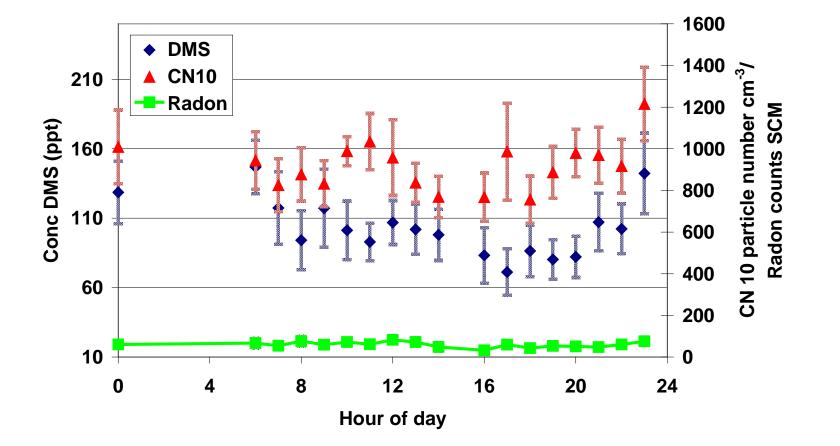
Phone: +61 3 9239 4428 Email: sarah.lawson@csiro.au Web: www.cawcr.gov.au

# Thank you





#### DMS diurnal cycle in clean air





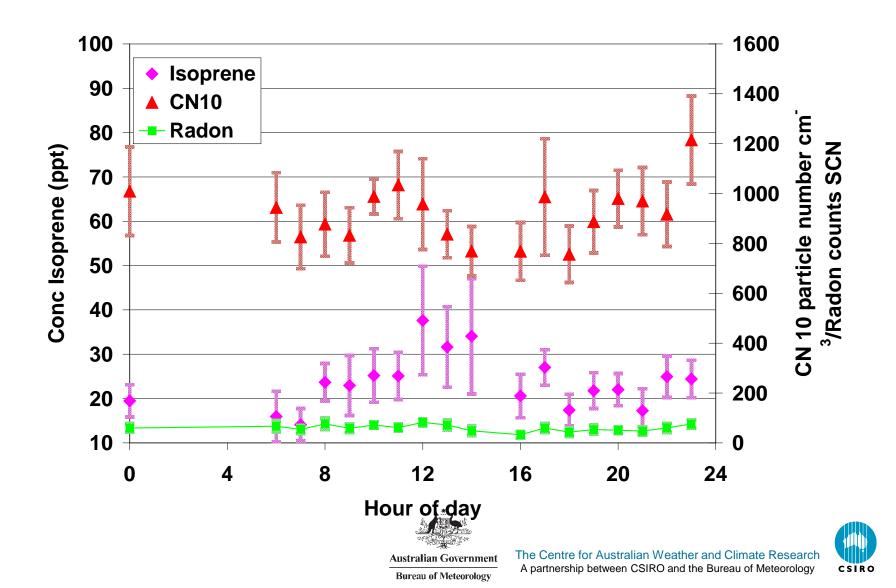
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#### Isoprene diurnal cycle clean air December



#### VOCs measured at Cape Grim ctd.



#### Medusa (2004 – ongoing)

benzene toluene acetylene ethylene ethane i-butane n-butane 1-3, butadiene i-pentane n-pentane isoprene ethylbenzene m+p-xylene o-xylene

C6H6 CH3C6H5 C2H2 C2H4 C2H6 *i*-C4H10 *n*-C4H10 C4H6 *i-*C5H12 *n*-C5H12 C5H8 C8H10 *m*+*p*-C8H10 o-C8H10

#### Medusa

There are preliminary calibration scales for ethane and benzene.

The other compounds are detected via primary and qualifying ions but the calibrations and data quality are not yet assessed

There are 7 background stations with Medusa instrumentation in the world.



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## VOC Measurement Programs at Cape Grim

- In-situ flask sampling and GC-MS analyses Nov Dec 1995 (Ye, Galbally, and Weeks, CSIRO)
- In-situ flask sampling and GC-FID analyses Nov 1998 Mar 2000 (Kivlighon, Galbally, and Weeks, CSIRO)
- Flask sampling on over-flights and GC-FID analyses Jan 1999 Mar 2000 (Kivlighon, Galbally, and Weeks, CSIRO)
- In-situ sampling and GC-FID analyses January & February 1999 (Lewis et al., University of Leeds)
- In-situ sampling and GC-MS analyses October 2004 ongoing (Krummel, Fraser, Steele, Porter, CSIRO, AGAGE)
- Air archive analyses and GC-MS analyses 1978 present (Krummel, Fraser, Steele, Porter, CSIRO, Miller SIO/NOAA)
- In-situ sampling and PTR-MS analyses 2006 ongoing on a campaign basis (Galbally, Lawson CSIRO)



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MDL



| Mass | VOC          | MDL for hourly<br>measurement<br>(ppt) | % measurements<br>> MDL (clean air) |
|------|--------------|--|-------------------------------------|
| 33   | Methanol     | 214                                    | 85%                                 |
| 42   | acetonitrile | 34                                     | 60%                                 |
| 45   | acetaldehyde | 27                                     | 65%                                 |
| 59   | acetone      | 31                                     | 60%                                 |
| 63   | DMS          | 28                                     | 90%                                 |
| 69   | isoprene     | 9                                      | 85%                                 |
| 81   | terpenes     | 6                                      | 90%                                 |

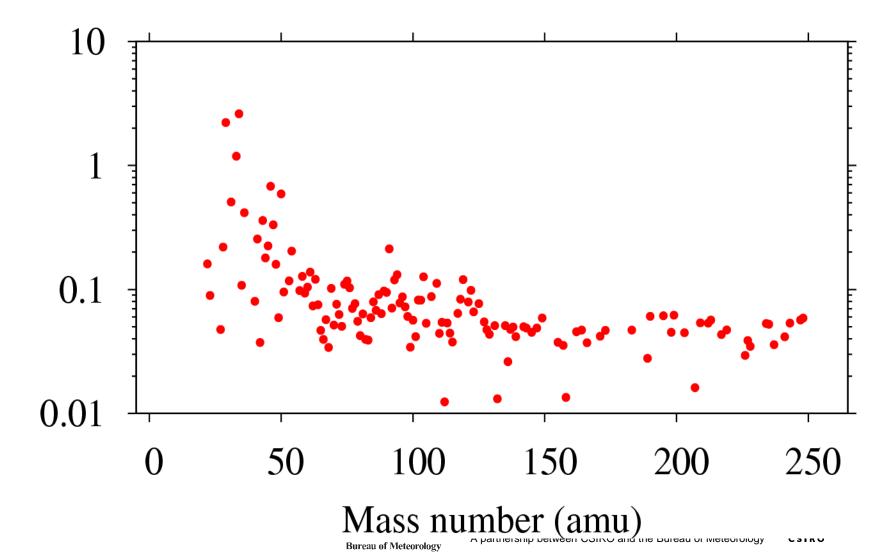
MDL calculation determined from scatter in zero measurements based on ISO6879





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#### Minimum Detectable Limit (MDL) vs Mass Number, for 1 sec dwell-time

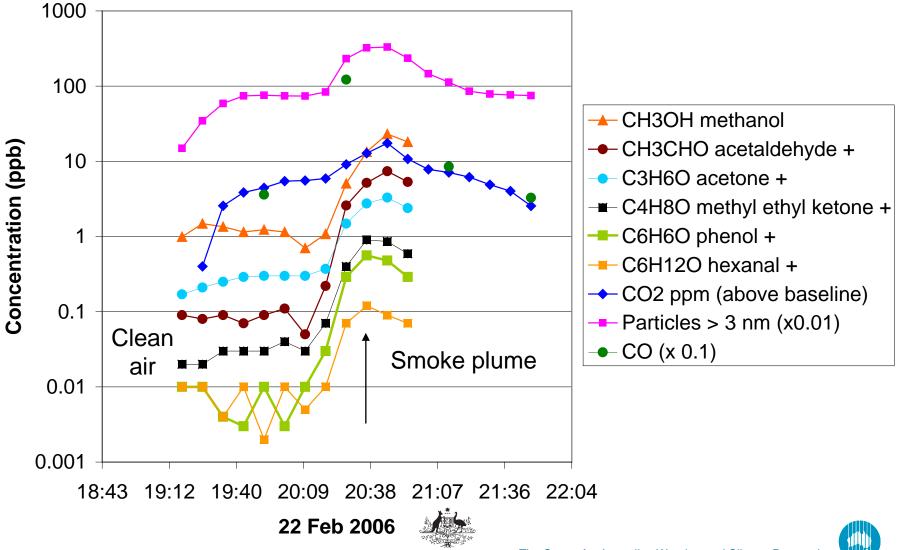


MDL (ppbv)

#### Oxygenated products



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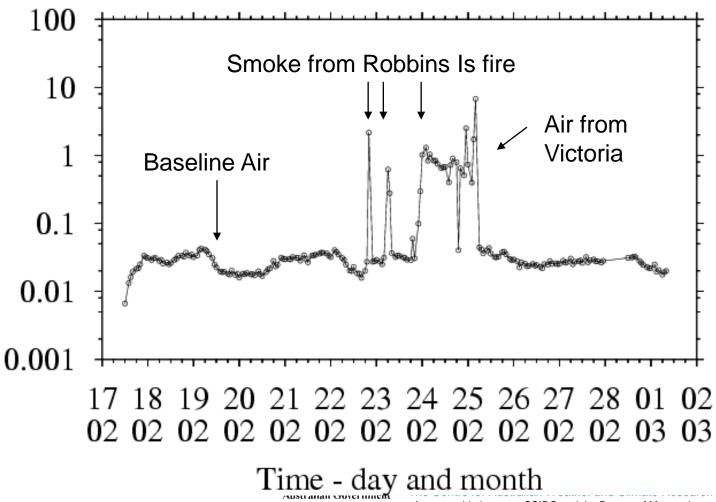
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#### Mass 42 Acetonitrile



#### Hourly concentrations - mass 42



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#### Baseline wind direction at Cape Grim

