

Measurement and Parameterization of Sea-Air Trace Gas Transfer using Micrometeorological Techniques A Decade of Progress



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Outline

- The need for air-sea gas transfer velocity (example for global CO₂)
- Wind speed parameterizations of k
- Direct observations of the flux (GasEx and more)
- COARE micro-meteorological *k* parameterization
- Addition of other gas flux observations into COARE
- Investigations of solubility and bubble effects
- Operational application of the COARE gas transfer parameterization





The need for k

20 10' Q¹ 10' 20* 30/ 401120 OP. 61.11 2009 Oct 21 10:47:45 -175 -135 -120 -105 -90 -75 -60 -45 -30 -15 0 15 30 45 60 75 90 105 120 135 168 -0.60-0.45-0.30-0.15 0.00 0.15 0.30 0.45 -1.75 1.75 ΔpCO₂ (Seawater-Air) (µatm) Total Flux (1012 grams C Month $\Delta pCO_2 * k$ **Flux**

Takahashi, et al. (2009), DSR II, 56, pp 554-577

ApCO₂ (Seawater-Air) (Rev Oct 09) for February 2000

k = ~ 0.26 U² Wanninkhof (1992)

MONTHLY Total Flux for February 2000 [Rev Oct 09] NCEP II Wind, 3,040K (U² wind, Γ=.26)

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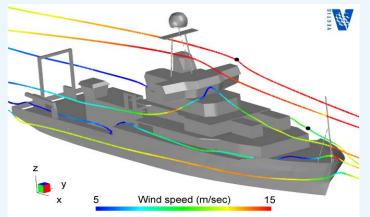
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Motion-Corrected Eddy-Covariance Turbulence Observations (momentum, sensible heat, and latent heat fluxes since 1990)









Physical flux: $\langle w'x' \rangle = C_x U(X_r - X_s) = C_x U\Delta X$ Gas flux: $\langle w'x' \rangle = k_x \alpha_x \Delta X$ α is solubility

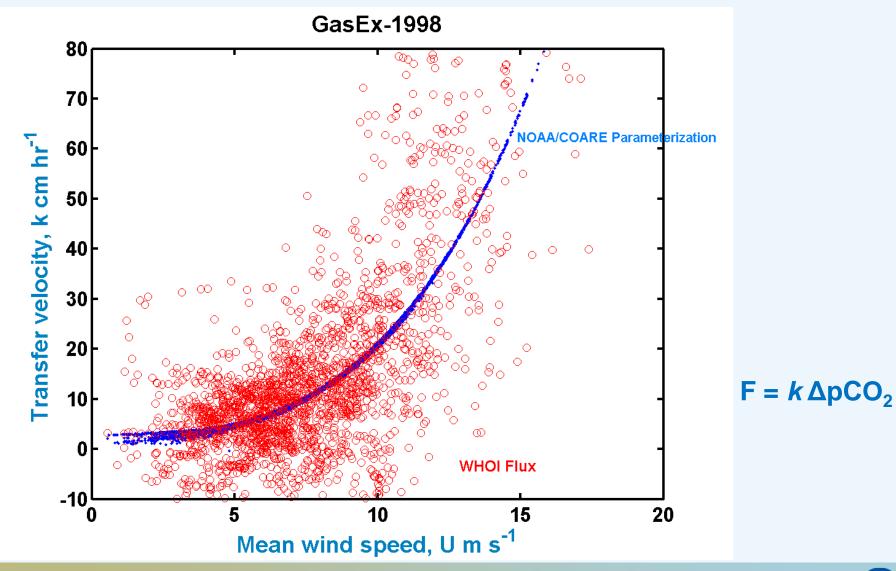
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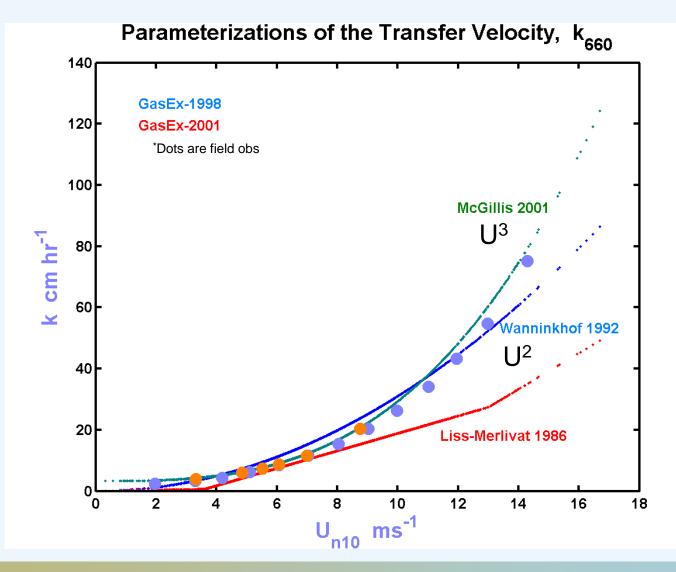
GasEx-98 cruise in the North Atlantic (~June 1998)



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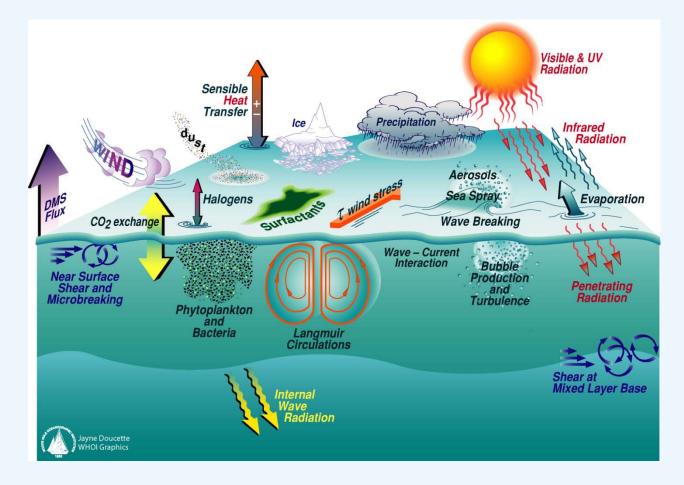


Quadratic, Cubic, relationships with U



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k varies only with U?



Development of a micro-meteorological

parameterization for gas transfer velocity, k

* Based on COARE bulk flux algorithm

 X_w = concentration in the water X_a = concentration in air

$$F_x = w'x' = \alpha_x k_x (X_{wr} / \alpha_x - X_{ar})$$

$$\alpha_{x} = X_{ws} / X_{as}$$

$$k_x = [R_{xw} + \alpha_x R_{xa}]^{-1}$$

Gases reactive in water Chemical enhancement factor β For large β , *k* becomes a deposition velocity

 $\alpha = \beta \alpha_x$ $F_x \rightarrow -\frac{X_{ar}}{R_{xa}} = -V_{dx} X_{ar}$





Expressions for resistances

Bubbles enhance transfer on ocean side

Air-side resistance Water-side resistance

A is adjustable constant ϕ is a buoyancy function

Bubble velocity (Woolf) B is adjustable constant W_b is whitecap fraction

$$k_{x} = [(R_{wx}^{-1} + k_{b})^{-1} + \alpha_{x}R_{ax}]^{-1}$$

 $u_{*a}R_{ax} = [h_a S_{ca}^{1/2} + C_d^{-1/2} - 5 + \ln(S_{ca}/(2\kappa))]$ $u_{*a}R_{wx} = \sqrt{\rho_w / \rho_a} [h_w S_{cw}^{1/2} + \ln(z_{wr}/\delta_{uw})]$

$$h_{w} = \frac{13.3}{A\phi}$$

Soloviev & Schlussel '94

$$k_{b} = \frac{V_{o}}{B} W_{b} \alpha_{x}^{-1} [1 + (e \alpha_{x} S_{cw}^{-1/2})^{-1/n}]^{-n}$$

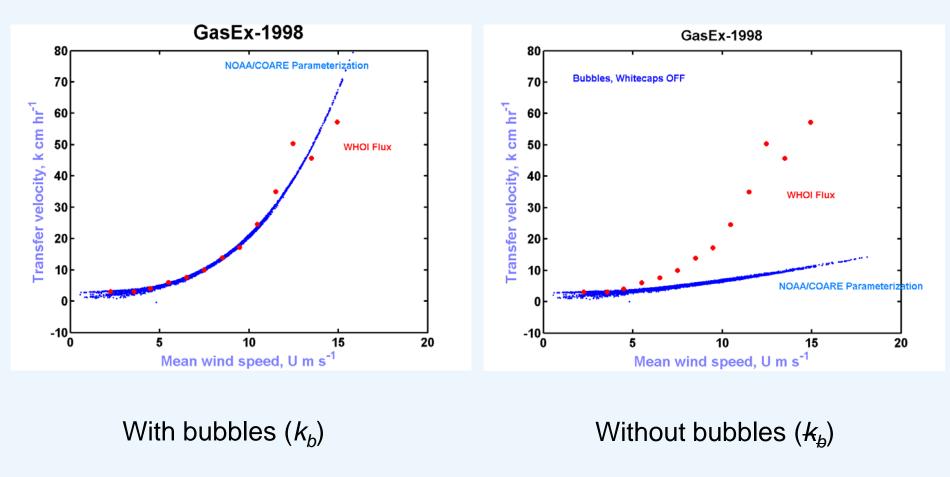
$$W_b = 3.8 * 10^{-6} U^{3.4}$$





Applied COARE gas parameterization

A and B adjustments

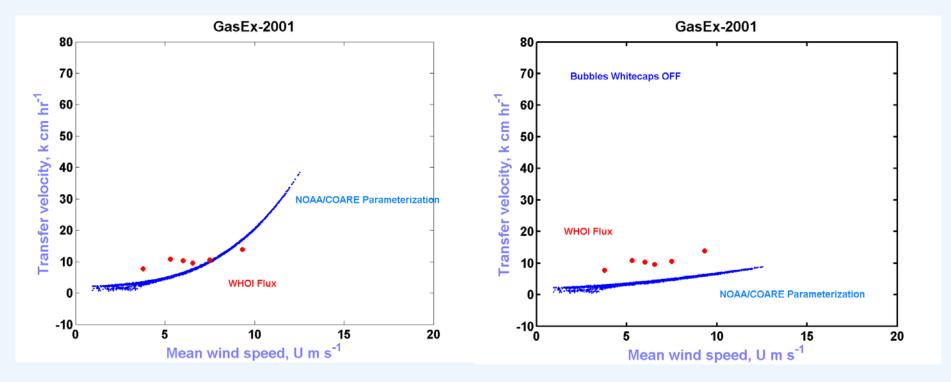


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GasEx-01 cruise in the Equatorial Pacific (~Feb 2001)



To match, A x3 B x2

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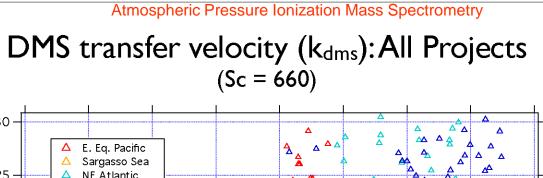
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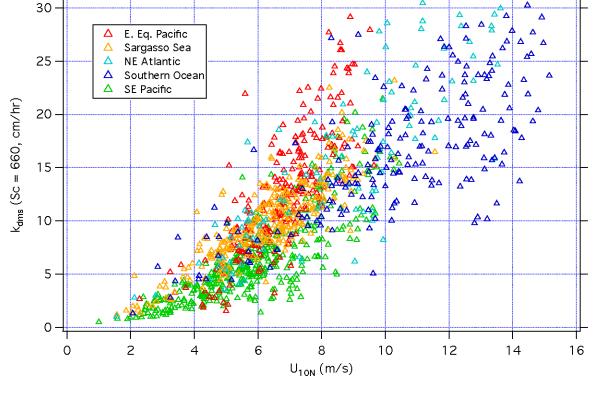




Adding more gases – DMS (CLAW)

Huebert and Blomquist - UHawaii



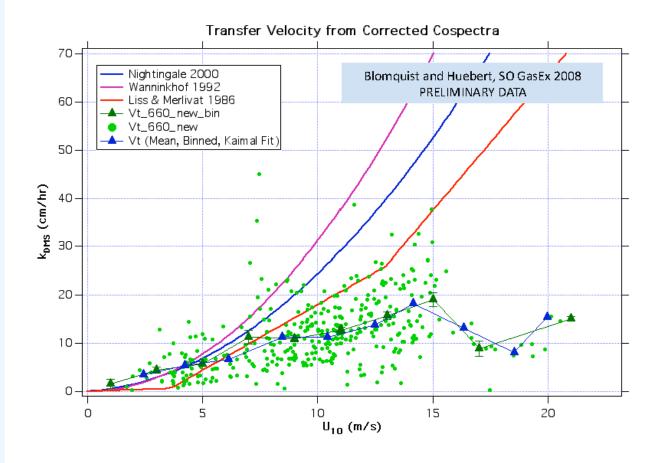


Increase data set, develop range of solubilities, elucidate the physics, etc

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Highlights the need to include solubility in k parameterization





Temperature effects of bubble-solubility normalization

15 -

$$\begin{bmatrix} \frac{S_{cw}}{660} \end{bmatrix}^{1/2} \frac{k_b}{u_*} \cong \frac{BV_o f_{wh}}{u_*} \frac{\gamma g_b}{\alpha (20)}$$

$$\gamma = \begin{bmatrix} \frac{S_{cw}}{660} \end{bmatrix}^{1/2} \frac{\alpha (20)}{\alpha (T)}$$

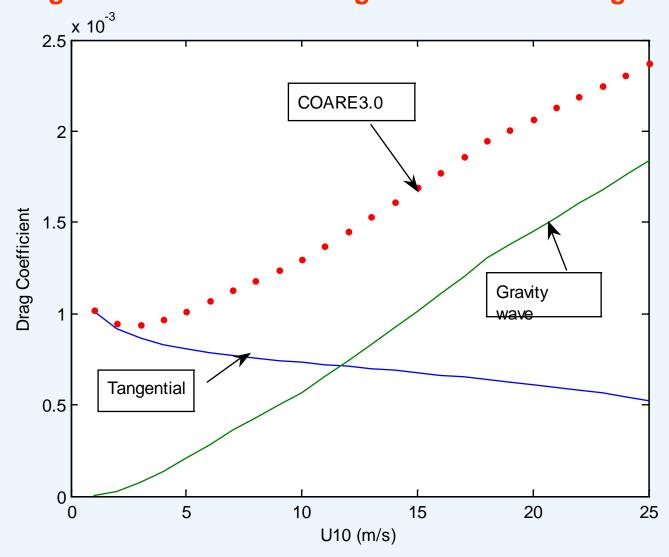
$$g_b = [1 + (e \frac{\alpha (20)}{\sqrt{660}} \gamma^{-1})^{-1/n}]^{-n}$$

How to discern bubble vs. interfacial transfer for field observations of the flux?





Partitioning of wind stress into tangential and form drag components

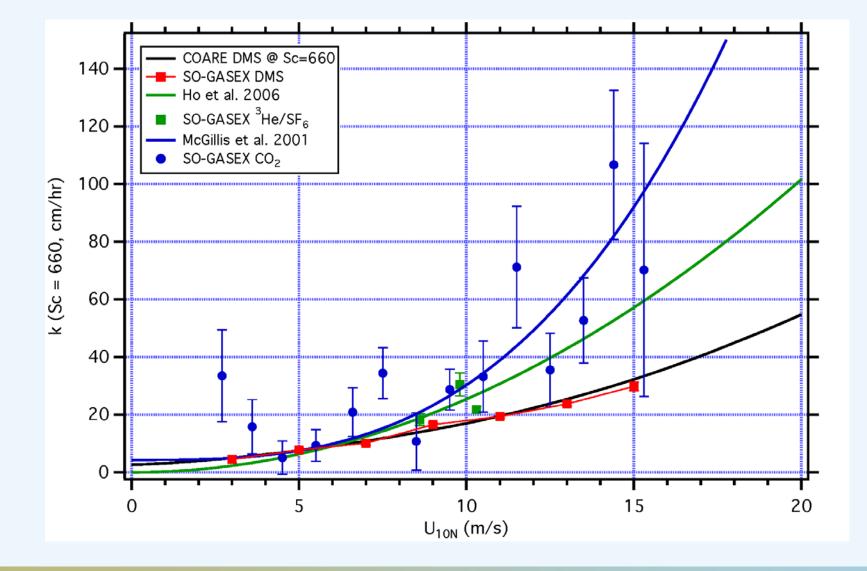


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GasEx-08 cruise in the Southern Ocean (~March 2008)





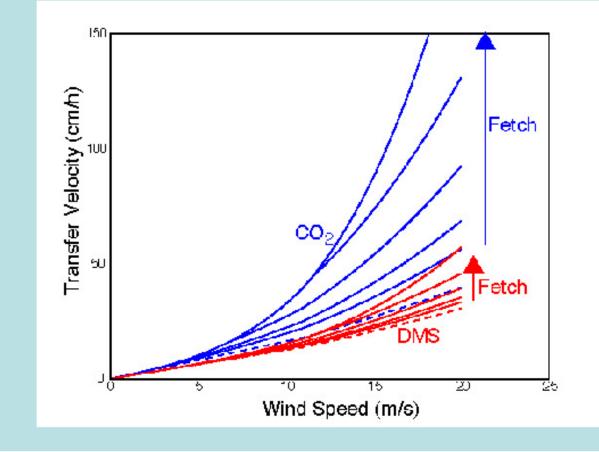






National Oceanography Centre, Southampton

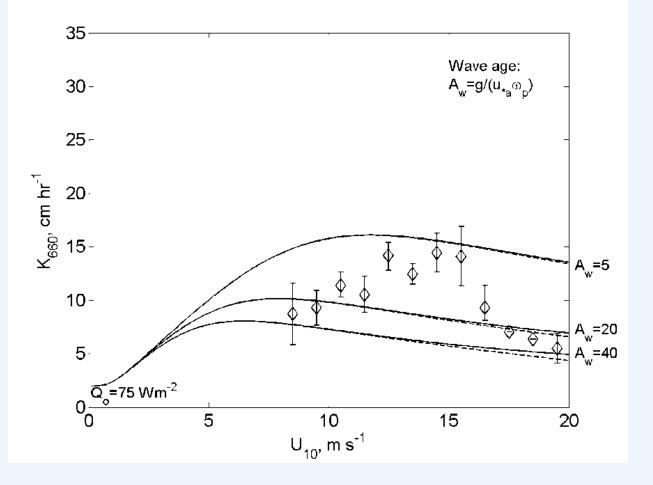
Woolf, D.K. 2006. Recent developments in parameterization of air-sea gas exchange. Flux News, 2,. In Press.







Wave state effects



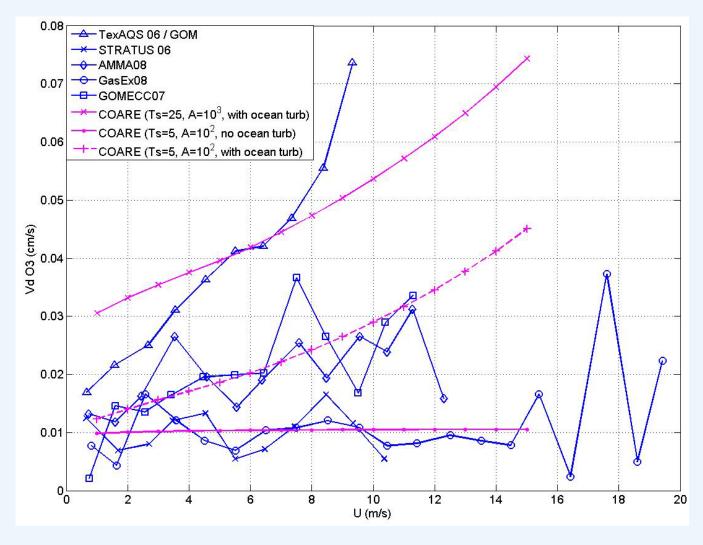
For a given wind speed, k drops for older waves (partitioning/breaking)





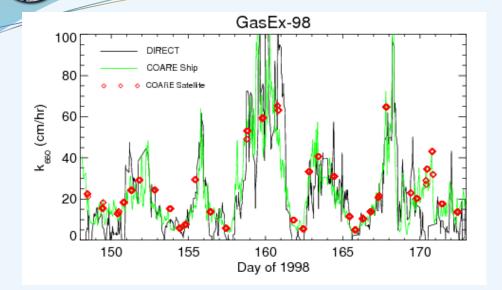
Adding more gases – O₃ (MBL and sfc chemistry)

Helmig – INSTAAR, University of Colorado O₃-NO-chemiluminescence



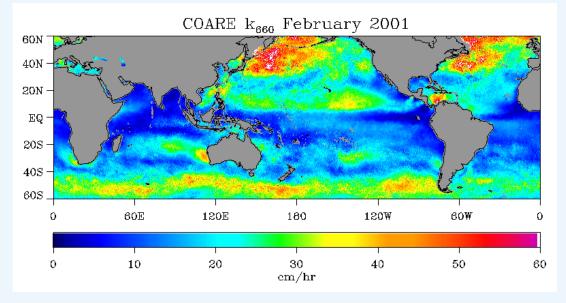


Satellite-derived CO₂ Transfer Velocities using COARE



COARE-modeled transfer velocities using satellite observations (*A*=1.31, *B*=2.57)

Monthly-mean k for CO₂ COARE with satellite inputs of U₁₀, T_{air}, T_{sea}, Q_{air}

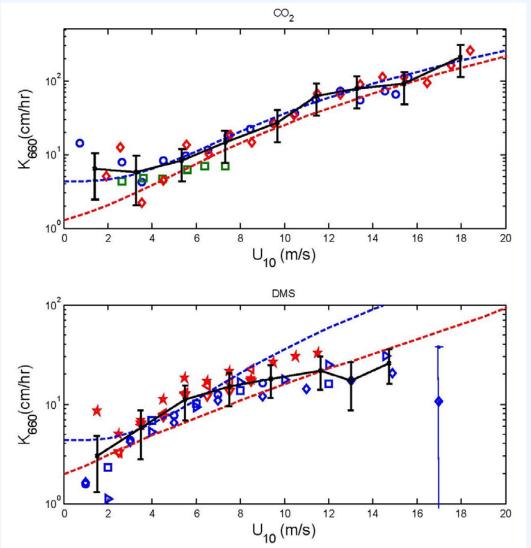


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Contrast to Stress/Heat Coefficients: Large

Uncertainties Remain for Gas Transfer



Gas Transfer Sensitivity to: *Solubility *Wave breaking *Bubbles *Tangential vs Pressure (wave) stress *Surfactants *Temperature *Complex chemistry *Biology





Summary

Progress on observational capabilities for air-sea gas flux observations
 Instrumentation
 Suite of gases

•Progress on development of comprehensive gas transfer algorithm

- •Gas transfer velocity determined at forcing scales
- Development needed for bubbles/whitecap range

Significant gaps exist in understanding gas exchange at high winds
Bubbles
Wave state

Understanding of surface and boundary layer biogeochemistry is needed
 Ozone deposition

